Chem 130 Fall 2016 Coastline College

Dupon

Homework Set 9

(Distributed 11/9/16; Due on 11/9/16)

Read Chapter 12 in Zumdahl and complete the listed questions from the text: 12, 17, 26, 34, 39, 46. 59, 65, 68, 78, 82; as well as the following problems:

A. Write the symbols for four ions that are isoelectronic with Xe.

 B. Indicate the direction of the polarity in each of the following bonds: C-Cl N-O H-O Which bond is expected to be the most polar? Why?

C. Draw Lewis dot structures for each of the following molecules:
(i) SBr₂
(ii) CO

(iii) $\underline{N}F_3$ (iv) $O\underline{C}Cl_2$

(v) $\underline{C}S_2$

D. Indicate the expected shape and bond angles around the underlined atom for each of the molecules in part C.